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=> fil caplus
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COST IN U.S. DOLLARS SINCE FILE TOTAL ENTRY SESSION FULL ESTIMATED COST 0.21 0.21

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FILE COVERS 1907 - 10 Nov 2008 VOL 149 ISS 20 FILE LAST UPDATED: 9 Nov 2008 (20081109/ED)

Caplus now includes complete International Patent Classification (IPC) reclassification data for the second quarter of 2008.

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http://www.cas.org/legal/infopolicy.html

=> s oxidation catalyst 492084 OXIDATION

820579 CATALYST

L1 1983 OXIDATION CATALYST (OXIDATION(W)CATALYST)

=> s 11 and tungsten 210737 TUNGSTEN

L2 58 L1 AND TUNGSTEN

=> s 12 and phosphoric acid 114184 PHOSPHORIC

4705251 ACID

104787 PHOSPHORIC ACID

(PHOSPHORIC (W) ACID)

L3 1 L2 AND PHOSPHORIC ACID

=> d bib abs

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ANSWER 1 OF 1 CAPLUS COPYRIGHT 2008 ACS on STN
T.3
    1987:578602 CAPLUS
ΑN
      107:178602
DN
OREF 107:28659a,28662a
TI Olefin oxidation catalyst system
      Vasilevskis, Janis; De Deken, Jacques C.; Saxton, Robert J.; Wentrcek,
ΙN
       Paul R.; Fellmann, Jere D.; Kipnis, Lyubov S.
PΑ
       Catalytica Associates, USA
SO
       PCT Int. Appl., 96 pp.
       CODEN: PIXXD2
DT
       Patent
LA
      English
      PATENT NO. KIND DATE APPLICATION NO.
WO 8701615 A1 19870336 WO 1001
FAN.CNT 1
                                                                                      19860918
PΤ
          W: AU, BR, DK, FI, JP, KR, NO
           RW: AT, BE, CH, DE, FR, GB, IT, LU, NL, SE
       US 4720474
                       A 19880119 US 1985-779501
                                                       US 1986-846554
IN 1986-CA651
                                                                                      19850924
      US 4723041 A 19880202 US 1986-846554
IN 168521 A1 19910420 IN 1986-CA651
ZA 8606653 A 19870729 ZA 1986-6653
AU 8663752 A 19870407 AU 1986-63752
EP 238633 A1 19870930 EP 1986-906113
                                                                                      19860331
                                                                                      19860828
                                                                                       19860902
                                                                                       19860918
                                                                                       19860918
           R: AT, BE, CH, DE, FR, GB, IT, LI, LU, NL, SE
BR 8606883 A 19871103 BR 1986-6883 19860918
JP 63500923 T 19880407 JP 1986-504984 19860918
IL 80091 A 19910630 IL 1986-80091 19860919
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CA 1268166 A1 19900424 CA 1986-518817 19860923
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DK 8702623 A 19870522 DK 1987-2623 19870522
FI 8702275 A 19870522 FI 1987-2275 19870522
NO 8702167 A 19870715 NO 1987-2167 19870522
US 4853357 A 19890801 US 1987-103442 19870930
PRAI US 1985-779501 A 19850924
WO 1986-US1950 A 19860918
OS CASREACT 107:178602
       BR 8606883 A 19871103 BR 1986-6883
                                                                                       19860918
OS
       CASREACT 107:178602
      Olefins are oxidized to carbonyl compds. in the presence of O and a
       catalyst system comprising a polyoxoanion and [XxMaM'bM''cOz]-m (M, M',
       M'' = W, Mo, V, Nb, Ta, Re; X = B, Si, Ge, P, As, Se, Te, I, Co, Mn, Cu;
       a, m, z = >0; b, c = integer; x = 0 for isopolyoxoanions, >0 for
       heteropolyoxoanions; such that a + b + c \ge 2), \ge 1 Pd
       component, ≥1 redox-active metal selected from CuSO4, Cu(OAc)2,
       Cu(NO3)2, Fe(OAc)2, FeSO4, and MnSO4, and a ligand. Thus, 73.2 g NaVO3
       was dissolved in 380 mL H2O which had been heated to 90° forming a
       first solution, which was added to a 90° solution consisting of 120 mL
       \rm H2O and 80.7 g \rm Na2MoO4.2H2O. To this mixture, 50 mL of 85% \rm H3PO4 was added
       dropwise, the solution heated to 95° for 1 h, filtered through Celite,
       .apprx.80 g K2SO4 was added to the filtrate which had been cooled to room
       temperature, the solution stirred for 1-1 1/2 h, and the K9PMo6V6O40
precipitate (I) was
       recrystd. from 0.25 M H2SO4. Li9PMo6V6O40 was prepared from I by
       ion-exchange chromatog., and served as the source for PMo6V6O40-9 (II).
```

recrystd. from 0.25 M H2SO4. Li9PMo6V6O40 was prepared from I by ion-exchange chromatog, and served as the source for PMo6V6O40-9 (II). 1-Hexene was oxidized in the presence of 15 mL H2O, 1.5 mL 1 normal H2SO4, 625 mg II, and a 1:5:10 molar ratio of Pd(CS3CO2)2-II-CuSO4.2H2O. The oxidation was accomplished at 85°/80 psig O2 for 8 h producing 73.7 mol% 1-hexene conversion with 95.0 mol% selectivity to 2-hexanone, vs. 53.7 mol% and 90.8 mol% resp., for a control oxidation conducted without CuSO4.

```
=> s 12 and hydrogen sulfate salt
       1096189 HYDROGEN
        570848 SULFATE
       865530 SALT
            52 HYDROGEN SULFATE SALT
                 (HYDROGEN (W) SULFATE (W) SALT)
             0 L2 AND HYDROGEN SULFATE SALT
L4
=> s 12 and quaternary ammonium salt
       143300 QUATERNARY
        433282 AMMONIUM
        865530 SALT
        10362 QUATERNARY AMMONIUM SALT
                 (QUATERNARY (W) AMMONIUM (W) SALT)
L_5
             0 L2 AND QUATERNARY AMMONIUM SALT
=> s 12 and boric acid
        46238 BORIC
       4705251 ACID
        43251 BORIC ACID
                 (BORIC(W)ACID)
L6
             1 L2 AND BORIC ACID
=> d bib abs
    ANSWER 1 OF 1 CAPLUS COPYRIGHT 2008 ACS on STN
1.6
ΑN
    2000:351422 CAPLUS
DN
    132:349279
ΤI
    Oxidation catalyst and process for producing oxole
    compound with the same
    Ito, Masumi; Sueyoshi, Tsuyoshi; Nakajima, Yasuko; Koyasu, Yukio
IN
    Mitsubishi Chemical Corporation, Japan
PA
SO
    PCT Int. Appl., 23 pp.
    CODEN: PIXXD2
DT
    Patent
LA
    Japanese
FAN.CNT 1
    PATENT NO.
                       KIND DATE
                                         APPLICATION NO.
                        ____
                               _____
                                          _____
                                         WO 1999-JP6362
PΤ
                        A1 20000525
                                                                 19991115
        W: CA, CN, IN, KR, SG, US
        RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL,
            PT, SE
    JP 2000210562
                               20000802
                                           JP 1999-323671
                        Α
PRAI JP 1998-324732
                       A
                              19981116
    Alkadienes are oxidized to oxoles in the presence of catalysts, WxA1-xOy,
    wherein A represents ≥1 element selected from alkali metals, Cr,
    Mo, Group 13 elements, and Group 15 elements other than Sb; 0 < x < 1; and
    y \neq 0, determined by the oxidation states of the other elements. Thus,
    1,3-butadiene (I) was oxidized at 438° and 1500 h-1 over
    W0.75B0.250x to prepare furan at 38% selectivity at I conversion 4.5%.
RE.CNT 12
             THERE ARE 12 CITED REFERENCES AVAILABLE FOR THIS RECORD
             ALL CITATIONS AVAILABLE IN THE RE FORMAT
=> s multicomponent oxidation catalyst
         42232 MULTICOMPONENT
        492084 OXIDATION
        820579 CATALYST
T.7
             O MULTICOMPONENT OXIDATION CATALYST
```

(MULTICOMPONENT (W) OXIDATION (W) CATALYST)

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=>

Executing the logoff script...

=> LOG Y

COST IN U.S. DOLLARS	SINCE FILE	TOTAL
	ENTRY	SESSION
FULL ESTIMATED COST	43.10	43.31
DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)	SINCE FILE	TOTAL
	ENTRY	SESSION
CA SUBSCRIBER PRICE	-1.60	-1.60

STN INTERNATIONAL LOGOFF AT 11:35:48 ON 10 NOV 2008